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#### **Key indicators**

Single-crystal X-ray study T = 273 K Mean  $\sigma$ (C–C) = 0.004 Å R factor = 0.053 wR factor = 0.133 Data-to-parameter ratio = 13.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# 5-Acetyl-4-methyl-2-(2-fluorophenylamino)-1,3-thiazole

The molecule of the title compound,  $C_{12}H_{11}N_2FOS$ , is essentially planar, with a maximum deviation of 0.127 (3) Å for the C atom of the methyl group attached to the thiazole ring. The structure is stabilized by intra- and intermolecular hydrogen bonds to form one-dimensional polymeric chains parallel to the *b* axis. Received 16 September 2005 Accepted 22 September 2005 Online 30 September 2005

## Comment

The title compound, (I), is isostructural with 5-acetyl-4methyl-2-(o-toluidinyl)-1,3-thiazole (Yamin *et al.*, 2005), except that the F atom is bonded at the *ortho* position of the phenylamino fragment. The whole molecule is essentially planar, with a maximum deviation for atom C12 of 0.127 (3) Å from the least-squares plane; in the molecule of 5-acetyl-4methyl-2-(o-toluidinyl)-1,3-thiazole, the benzene group is inclined to the thiazole ring by 73.44 (10)°. All bond lengths and angles are in normal ranges (Allen *et al.*, 1987) and are comparable with those in the analogue.



There are weak intramolecular N1-H1···F1, C10-H10B···O1 and C5-H5···N2 hydrogen bonds (Table 2). In the crystal structure, the molecules are linked by inter-



© 2005 International Union of Crystallography Printed in Great Britain – all rights reserved Figure 1 The molecular structure of (I), with 50% probability displacement ellipsoids.



### Figure 2

A packing diagram for (I), viewed down the b axis. Dashed lines denote the N-H···O and C-H···F hydrogen bonds.

molecular N1-H1···O1<sup>i</sup> and C12-H12C···F1<sup>ii</sup> hydrogen bonds (symmetry codes as in Table 2) to form one-dimensional zigzag polymeric chains parallel to the *b* axis (Fig. 2).

## **Experimental**

A solution of o-fluoroaniline (1.11 g, 0.01 mol) in acetone (50 ml) was added dropwise to an acetone solution (50 ml) containing an equimolar amount of 3-chloroacetylacetone and ammonium thiocyanate in a two-necked round-bottomed flask. The solution was refluxed for about 2 h. The light-yellow solution was filtered off and colourless crystals of (I) were obtained after evaporation over a period of 5 d (yield 85%; m.p. 447.8-449.3 K).

## Crystal data C12H11FN2OS

 $M_r = 250.29$ 

Monoclinic, $P2_1/c$	Cell pa
a = 5.0205 (11)  Å	refle
b = 13.081 (3) Å	$\theta = 1.9$
c = 17.598 (4) Å	$\mu = 0.2$
$\beta = 97.085 \ (4)^{\circ}$	T = 273
V = 1146.9 (4) Å <sup>3</sup>	Block,
Z = 4	0.43 $\times$
Data collection	
Bruker SMART APEX CCD area-	2115 in
detector diffractometer	1931 re
$\omega$ scans	$R_{\rm int} = 0$
Absorption correction: multi-scan	$\theta_{\rm max} = 1$
(SADABS; Bruker, 2000)	h = -6
$T_{\min} = 0.889, T_{\max} = 0.954$	k = -1
6015 measured reflections	l = -13

 $D_{\rm r} = 1.450 {\rm Mg m}^{-3}$ Mo  $K\alpha$  radiation Cell parameters from 2840 ections -26.5°  $28 \text{ mm}^{-1}$ 3 (2) K colourless  $0.28 \times 0.17 \text{ mm}$ 

2115 independent reflections
1931 reflections with $I > 2\sigma(I)$
$R_{\rm int} = 0.018$
$\theta_{\rm max} = 25.5^{\circ}$
$h = -6 \rightarrow 5$
$k = -15 \rightarrow 13$
$l = -13 \rightarrow 21$

#### Refinement

2	2 2 2
Refinement on $F^2$	$w = 1/[\sigma^2(F_0^2) + (0.0582P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.053$	+ 0.5434P]
$wR(F^2) = 0.133$	where $P = (F_0^2 + 2F_c^2)/3$
S = 1.19	$(\Delta/\sigma)_{\rm max} < 0.001$
2115 reflections	$\Delta \rho_{\rm max} = 0.23 \ {\rm e} \ {\rm \AA}^{-3}$
160 parameters	$\Delta \rho_{\rm min} = -0.19 \text{ e} \text{ \AA}^{-3}$
H atoms treated by a mixture of	
independent and constrained	
refinement	

Table 1

Selected geometric parameters (Å, °).

\$1-C7	1.733 (3)	N1-C6	1.401 (3)
S1-C9	1.740 (2)	N2-C7	1.309 (3)
F1-C1	1.358 (3)	N2-C8	1.375 (3)
O1-C11	1.214 (3)	C8-C9	1.370 (4)
N1-C7	1.360 (3)		
C7-S1-C9	88.76 (11)	C7-N2-C8	110.8 (2)

Table 2			
Hydrogen-bond geometry	(Å,	°).	

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdot \cdot \cdot A$
$N1 - H1 \cdots F1$	0.87 (3)	2.27 (3)	2.653 (3)	107 (2)
$C5-H5\cdots N2$	0.93	2.30	2.919 (4)	123
C10−H10B···O1	0.96	2.40	3.018 (4)	122
$N1 - H1 \cdots O1^{i}$	0.87 (3)	2.18 (3)	3.010 (3)	158 (3)
$C12-H12C\cdots F1^{ii}$	0.96	2.33	3.165 (3)	144

Symmetry codes: (i)  $-x, y - \frac{1}{2}, -z + \frac{1}{2}$ ; (ii)  $-x, y + \frac{1}{2}, -z + \frac{1}{2}$ .

All H atoms were located in a difference map. The H atom on N1 was refined isotropically. All other H atoms were placed geometrically in ideal positions and allowed to ride on their parent C atoms, with C-H distances in the range 0.93–0.96 Å and with  $U_{iso}(H) =$  $1.5U_{eq}$  (methyl C), or  $1.2U_{eq}$  (C) for CH groups.

Data collection: SMART (Bruker, 2000); cell refinement: SAINT (Bruker, 2000); data reduction: SAINT; program(s) used to solve structure: SHELXTL (Sheldrick, 1997); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL, PARST (Nardelli, 1995) and PLATON (Spek, 2003).

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